



INDIAN INSTITUTE OF TECHNOLOGY GUWAHATI
SHORT ABSTRACT OF THESIS

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Programme of Study : Ph.D.

Thesis Title: "**Synthesis, Characterization and Spectroscopic Studies of Transition Metal Complexes with Chalcogen (O, S, Se and Te) Bridged Non-innocent Ligands**"

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SHORT ABSTRACT

Amidophenolate based non-innocent ligands and their corresponding transition metal (M) complexes are very effective for enzymic mimicking, catalytic C–C cross coupling reaction, –CF₃ group transferring agent, C–O/N/S bond activation, valence tautomeric species formation, synthesis of M–quinone species, etc. Thus, a few non-innocent ligands with two orthogonally placed aminophenol units were designed and synthesized by placing a common chalcogen (O, S, Se and Te) bridge between two H₂L^{AP} [N(2-hydroxy-3,5-di-*tert*-butyl phenyl) aniline] ligand units. The sulfur bridged ligand H₄L^{S(AP/AP)} provided an unprecedented octahedral complex, **2E**, where one of the two coordinated amidophenolate units was present in its two electron oxidized iminoquinone {[IBQ]⁰} form and other was in its one-electron oxidized iminosemiquinone {[ISQ]⁻¹} form. The oxygen bridged ligand H₄L^{O(AP/AP)} provided a mononuclear octahedral cobalt complex **3A** by two set of successive homolytic C_{aryl}–O_{bridge} bond breaking and homolytic Co–O_{bridge} bond formation. Complex **3A** showed solvent and solid phase thermal tautomerization and the solid phase tautomeric conversion depends upon the availability of the lattice solvent within the crystalline matrix. An unprecedented one-step synthesis of Cu(II)-bis(iminoquinone) complex using amidophenolate based ligand has been established by using non-innocent ligand H₂L^{AP(o-NO₂-OPh)} and the Cu(II)-complex mimicked the function of coenzyme ubiquinone. A hemilabile –NH₂ group appended aminophenol based ligand H₂L^{AP(o-NH₂-OPh)} was also designed and synthesized by reduction of ligand H₂L^{AP(o-NO₂-OPh)}. This ligand provided a square pyramidal Cu(II)-bis(iminosemiquinone) complex (**5D**), which possessed two closely located doublet spin states. Thus, the extent of coupling between Cu(II) and radical has been explored.